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42 For some hypothetical metal, the equilibrium number of vacancies at 900 C is $23 \times 10^{25} \text{ m}^{-3}$ If the density and atomic weight of this metal are 740 g/cm³ and 855 g/mol, respectively, calculate the fraction of vacancies for this metal at 900 C Solution This problem is solved using two steps: (1) calculate the total number of lattice sites in silver, N, using

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